

**Bureau de Normalisation des Liants Hydrauliques**

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**Certificate of Analysis**

**Reference Material SN1**  
**Portland Cement**  
**(CEM I 52,5 N)**

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**I – Participation and execution of tests**

Each year the “*Association Technique de l’Industrie des Liants Hydrauliques*” (ATILH) organises an interlaboratory test campaign involving in particular the participation of the cement production industry laboratories, the cement end-user laboratories and Research and Inspection Centers within the construction materials sector. This participation is compulsory for laboratories accredited by COFRAC for cement testing. The tests are carried out in accordance with standardised methods where latter exist, otherwise according to everyday traditional methods.

**II – Statistical analysis of the results**

Outliers are eliminated via the STUDENT’s test with a confidence level of 98 %. A reiteration is set at this threshold in order to keep only those values which are related to the “Normal or Gaussian” distribution, the latter being entirely defined by 2 parameters: mean and standard deviation. The coefficient of variation symbolised by “V” is the ratio between the standard deviation “ $\sigma$ ” and the mean value  $\bar{X}$  .

**III – Specific surface and particle size analysis**

For the calibration of the Blaine permeability apparatus, follow the requirements of the EN 196-6 standard, paying particular attention to the temperature corrections, if any. To determine the volume of the compacted layer, it is not essential to use the Reference Material (but ensure that a sufficient quantity is taken so that the mass of the mercury does not modify the compaction of the powder layer). Reference Material should be used systematically:

- a) after 1000 tests ;
- b) when using another type of manometric liquid, another type of filter paper, a new manometer tube or a new perforated disc;
- c) If discrepancies are systematic with the secondary reference cement.

Table 1

	<b>Mean value</b>	<b>Dispersion characteristics Reproducibility</b>	
		$\sigma$	V (%)
Particle density (g/cm <sup>3</sup> ) with picnometer method	<b>3,11</b>	<b>0,02</b>	<b>0,6</b>
Blaine Specific Area (cm <sup>2</sup> /g) with EN 196-6	<b>3396</b>	<b>92</b>	<b>2,7</b>

Table 2

Particle size analysis by laser diffraction (ISO 13320-1)			Air-jet sieving - Alpine test (NF X11-640)	
Equivalent size aperture a ( $\mu\text{m}$ )	Mean (% of < a)	$\sigma$ (%) □(reproducibility)□	Mean (% of < a)	$\sigma$ (%) □(reproducibility)□
2,0	11,4	3,7		
3,15	16,8	4,2		
4	20,1	4,1		
5	23,3	4,1		
6,3	26,9	4,0		
8	31,0	3,8		
12,5	40,4	3,9		
16	46,4	4,6		
25	60,7	4,9		
31,5	68,6	5,1	71,4	6,8
40	77,9	5,0	82,3	2,1
50	85,7	4,6	88,6	2,0
63	92,0	3,5	93,8	1,8
80	96,0	2,2	97,8	1,4
100	98,6	1,2	99,0	1,0
125	99,5	0,7	99,5	0,5
160	99,9	0,3	100	0,04

### III – Chemical composition

X-ray fluorescence spectrometry, fused bead (ISO 29581-2)				Chemical Analysis (EN 196-2)		
Elements	Mean $\bar{X}$ (%)	Standard deviation $\sigma$ (%) reproducibility	Coefficient of variation V (%)	Mean $\bar{X}$ (%)	Standard deviation $\sigma$ (%) reproducibility	Coefficient of variation V (%)
Loss on ignition	-	-	-	1,39	0,09	6,12
SiO <sub>2</sub>	20,23	0,15	0,76	20,18	0,26	1,29
Al <sub>2</sub> O <sub>3</sub>	5,24	0,09	1,72	5,32	0,1	1,79
Fe <sub>2</sub> O <sub>3</sub>	2,00	0,04	1,77	2,03	0,09	4,21
CaO	65,77	0,29	0,45	65,72	0,39	0,6
MgO	1,13	0,04	3,85	1,15	0,1	8,4
SO <sub>3</sub> <sup>3</sup>	3,06	0,06	1,87	3,07	0,1	3,31
Free CaO <sup>2</sup>				0,83	0,22	27
Insoluble <sup>3</sup>				0,21	0,09	45
Na <sub>2</sub> O <sup>1</sup>	0,19	0,03	16,14	0,18	0,02	13
K <sub>2</sub> O <sup>1</sup>	0,28	0,01	5,06	0,28	0,02	8,65
TiO <sub>2</sub>	0,20	0,01	4,23			
P <sub>2</sub> O <sub>5</sub>	0,57*					
SrO	0,05*					

<sup>1</sup> photometric method<sup>3</sup> gravimetric method<sup>2</sup> all methods combined \* P<sub>2</sub>O<sub>5</sub> ±0,01 % - SrO ±0,004 %

### III – Sample conditioning

The sample of this reference material is packaged in 40 g glass bottle, sealed with a secure screw cap. Physico-chemical properties of the sample are stable until the bottle is closed and the cap untouched. After opening the bottle the local conditions of storage of the sample (courtroom with low humidity, maintaining in a drier, close the bottle immediately after use) will allow its potential reuse.